Our scientific workshop





Water tests in Plougastel-Daoulas, the Elorn river and Brest harbour November 2013

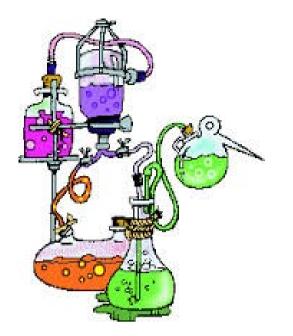


THE CONTEXT





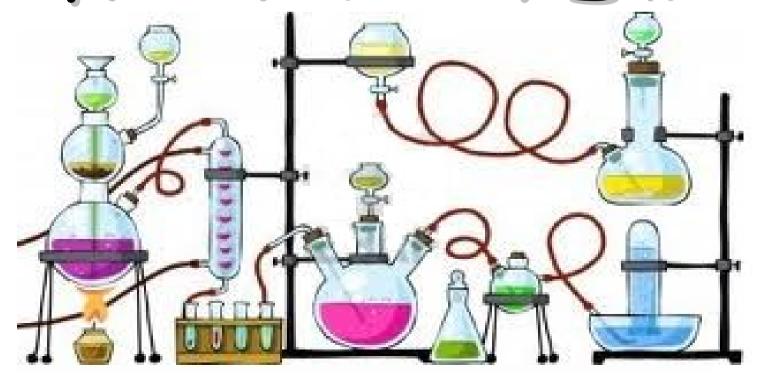
Within the frame of our COMENIUS project « Mare Nostrum » and in close relation with the Physics curriculum, pupils in year 10 have analysed water samples coming from different local places. Then, we have measured the concentration in nitrite, nitrat, ammonium and phosphat ions. Finally, we've measured the pH to controle the quality of these waters.







The Chemistry Lab





The tests were made in the Physics lab. Like scientists, we had to wear a white coat, gloves and glasses.



Sylvain

We split into 5 groups.

Each group had to test one ion or the pH.



Ewen We split into 5 groups.

Sylvain

Each group had to test one ion or the pH.



We wrote all the results of the tests on the board.

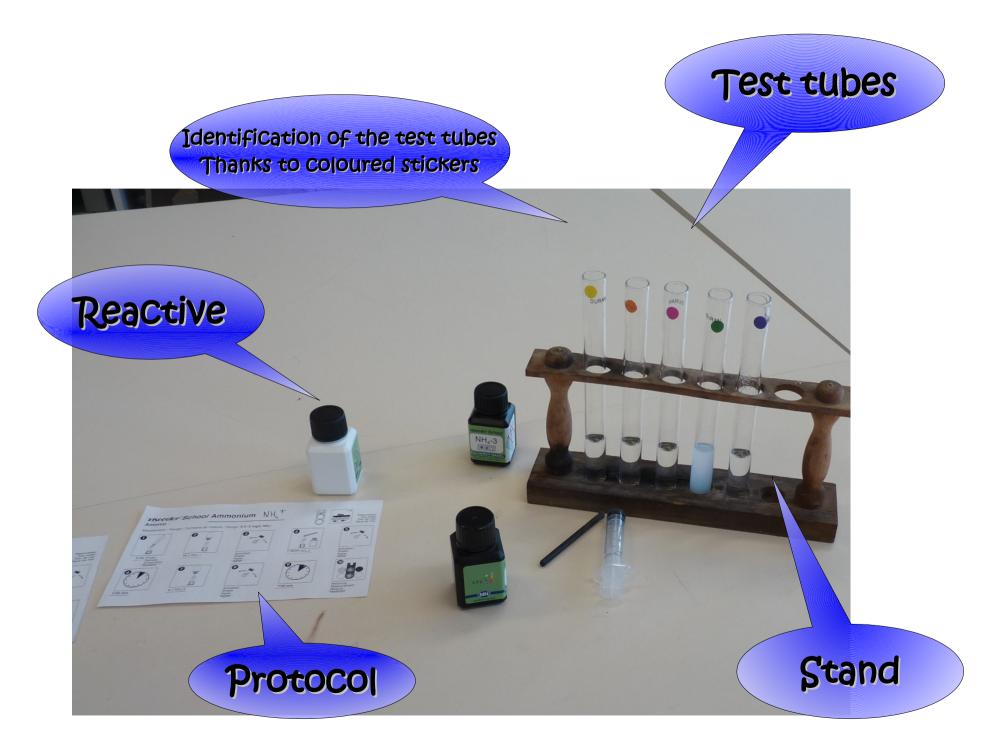


The scientific equipment



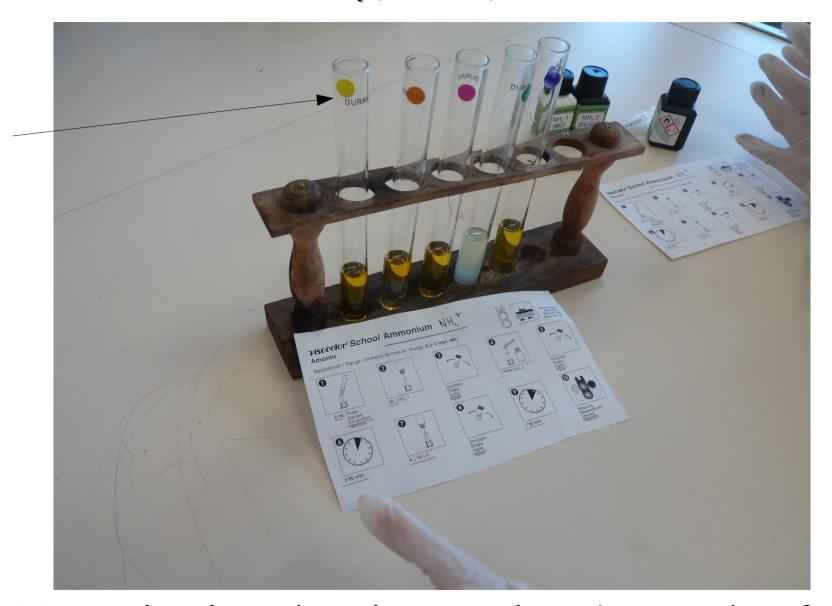


To do the tests we used specific chemistry materials

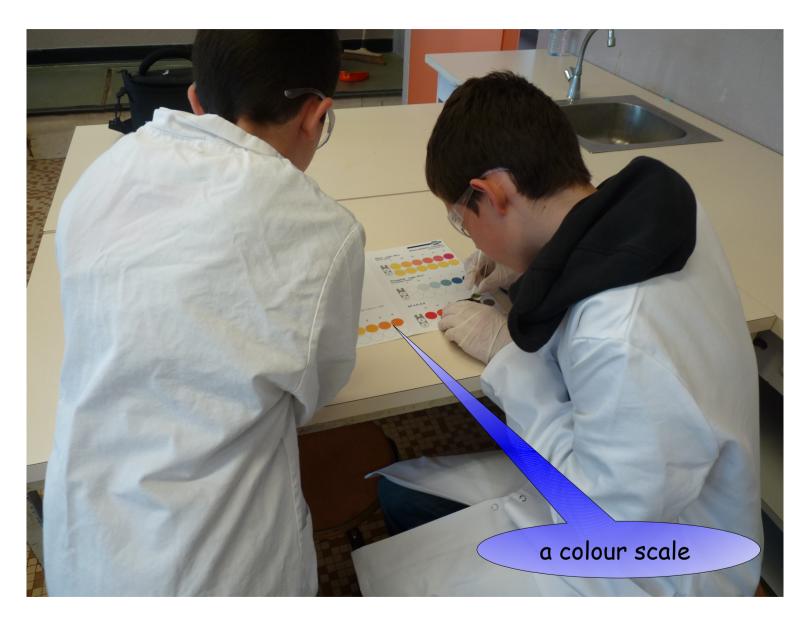


We had to follow a strict protocole to test the water samples

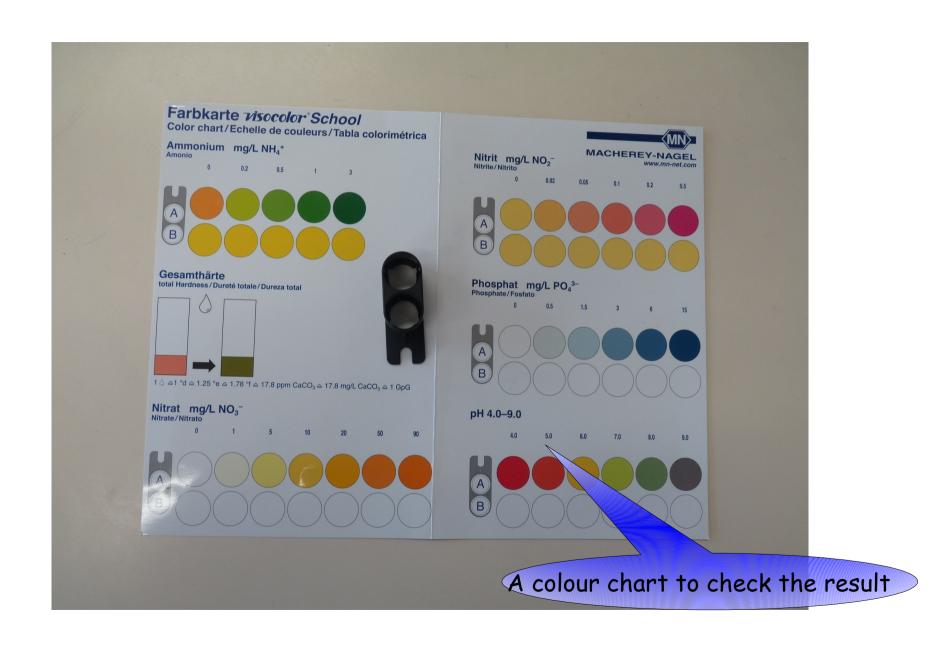
The tools



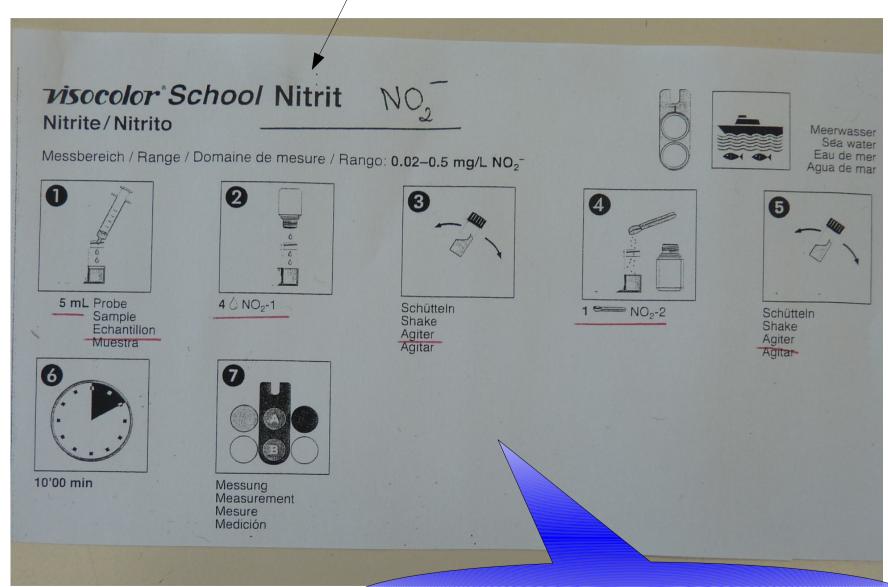
We stuck coloured stickers on the tubes to identify the samples



We also used a colour scale to measure the chemical substances concentration

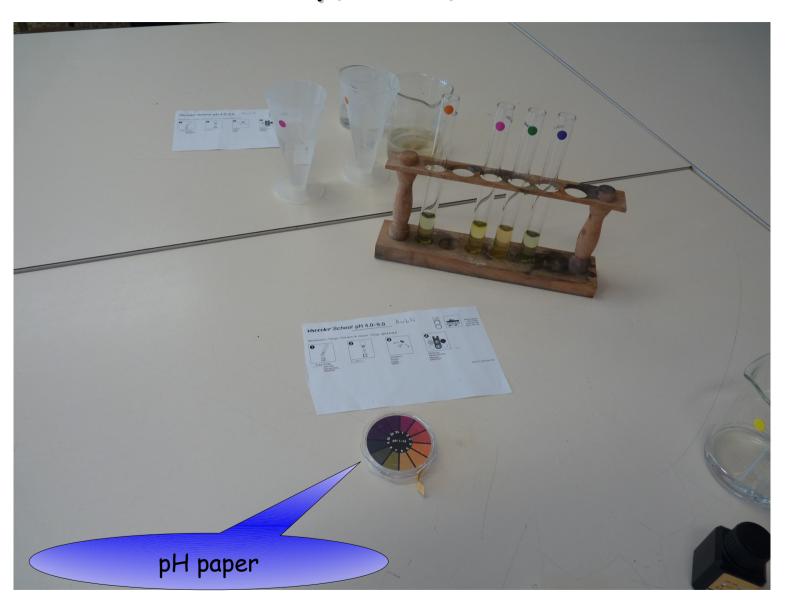


Here is the colour scale

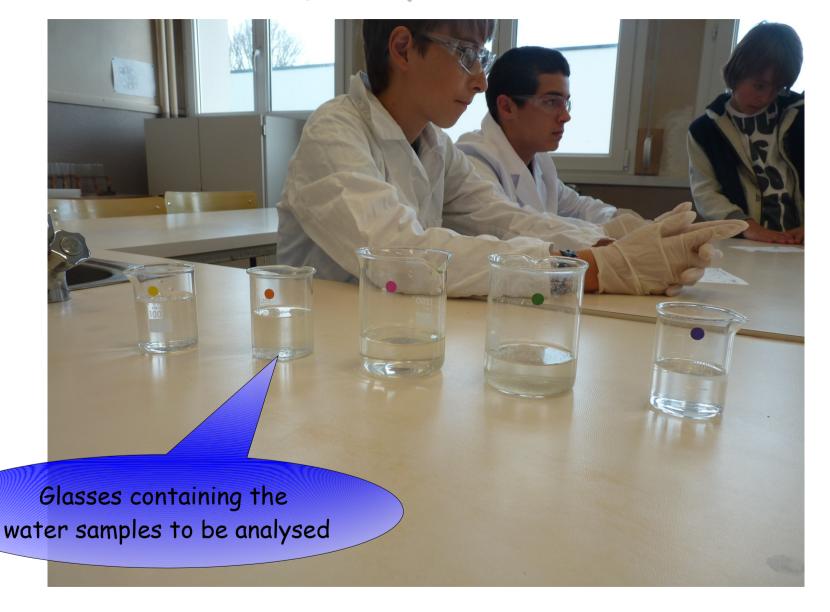


This is an example of an experimental protocol

The tools



The equipment



We took 5mL from each sample with a plastic syringe



Let's do the tests!





Nitrite ions tests



We added 4 drops of NO2-1 in each sample

Nitrite ions tests









Marianne

Lucie

Mr Héronneau Technology teacher

ENSEMBLE 50 ANS DE PROTECTION

JUPITER SATURNE SATURNE





pH test



Quentin

Maël

Mrs Lafont Biology teacher

pH test



Nitrite ions tests



Nitrite ions tests

Andrès



Valentin

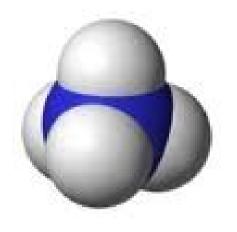
Colour scale measurement



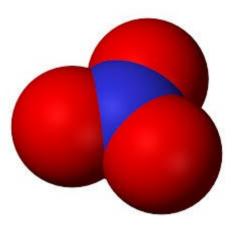
Phosphat ions tests

Mrs Karpus Physics teacher

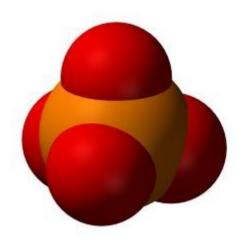
Sylvain



NH₄⁺



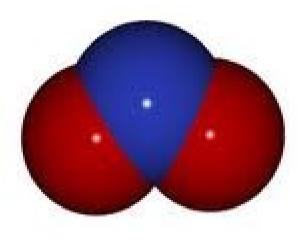
NO₃



PO₄³⁻

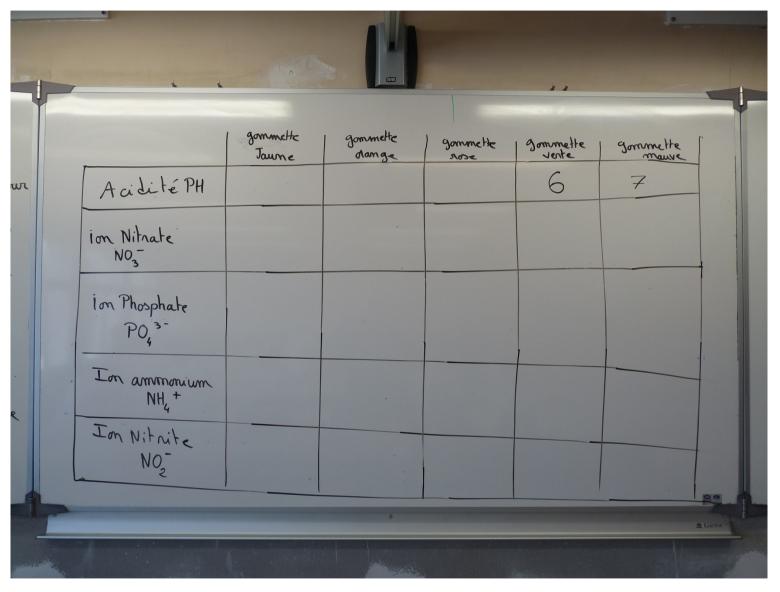


pH



NO₂

Here are the results we've obtained



One by one we wrote our results in the chart

				,,	
	gommette Jaune	gommette dange	gommette soose	Jomnmette Vente	gonumette maure
Aadité	PH 7	7	6	6	7
ion Nitrate	10 mg /L				5 mg/L
Ion Phosphate					-
Ion ammonia NH ₄ +					
Ion Nitrite NO2		0,05			





	0	SIIS		3 3	
	gommette Jaune	gommelte dange	gommette nose	gommelte vente	gommette maure
Acidité PH	F	7	6	6	7
ion Nitrate NO3	10 mg /L	5 mg/L	10 mg/L	1mg/L	5 mg/L
ion Phosphate PO ₄ 3-	1,5 mg/L	0,5 mg/L	0,5 mg/2	0,5 mg/L	1,5 mg/L.
Ion ammonium NH4+	0 mg/L	Orma IL	Omg 1	0,2mg/L	OmgK
Ion Nitrite NO2	0,2 mg))	0,05 mg/L	0,05 mg)	O,1mg/	Om9/2
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Many thanks to all the pupils who took part into this very interesting experiment: Lucie, Marianne, Quentin T, Yohann, Maël, Quentin D., Sylvain, Andrès, Valentin, Téo, Ewen, Costa et Mathys.

Year 8 Embassador pupils



Mathis

Costa

Year 10 Embassador Pupils



Téo

Ewen

This week, January 19th to 24th 2014, we are proud to be our school ambassadors





in Constanta, Romania!